International Journal of Physical Sciences

Volume 11 Number 20 30 October, 2016

ISSN 1992-1950



ABOUT IJPS

The International Journal of Physical Sciences (IJPS) is published weekly (one volume per year) by Academic Journals.

International Journal of Physical Sciences (IJPS) is an open access journal that publishes high-quality solicited and unsolicited articles, in English, in all Physics and chemistry including artificial intelligence, neural processing, nuclear and particle physics, geophysics, physics in medicine and biology, plasma physics, semiconductor science and technology, wireless and optical communications, materials science, energy and fuels, environmental science and technology, combinatorial chemistry, natural products, molecular therapeutics, geochemistry, cement and concrete research, metallurgy, crystallography and computer-aided materials design. All articles published in IJPS are peer-reviewed.

Contact Us

Editorial Office:	ijps@academicjournals.org		
Help Desk:	helpdesk@academicjournals.org		
Website:	http://www.academicjournals.org/journal/IJPS		
Submit manuscript online	http://ms.academicjournals.me/		

Editors

Prof. Sanjay Misra

Department of Computer Engineering, School of Information and Communication Technology Federal University of Technology, Minna, Nigeria.

Prof. Songjun Li

School of Materials Science and Engineering, Jiangsu University, Zhenjiang, China

Dr. G. Suresh Kumar

Senior Scientist and Head Biophysical Chemistry Division Indian Institute of Chemical Biology (IICB)(CSIR, Govt. of India), Kolkata 700 032, INDIA.

Dr. 'Remi Adewumi Oluyinka

Senior Lecturer, School of Computer Science Westville Campus University of KwaZulu-Natal Private Bag X54001 Durban 4000 South Africa.

Prof. Hyo Choi

Graduate School Gangneung-Wonju National University Gangneung, Gangwondo 210-702, Korea

Prof. Kui Yu Zhang

Laboratoire de Microscopies et d'Etude de Nanostructures (LMEN) Département de Physique, Université de Reims, B.P. 1039. 51687, Reims cedex, France.

Prof. R. Vittal

Research Professor, Department of Chemistry and Molecular Engineering Korea University, Seoul 136-701, Korea.

Prof Mohamed Bououdina

Director of the Nanotechnology Centre University of Bahrain PO Box 32038, Kingdom of Bahrain

Prof. Geoffrey Mitchell

School of Mathematics, Meteorology and Physics Centre for Advanced Microscopy University of Reading Whiteknights, Reading RG6 6AF United Kingdom.

Prof. Xiao-Li Yang

School of Civil Engineering, Central South University, Hunan 410075, China

Dr. Sushil Kumar

Geophysics Group, Wadia Institute of Himalayan Geology, P.B. No. 74 Dehra Dun - 248001(UC) India.

Prof. Suleyman KORKUT

Duzce University Faculty of Forestry Department of Forest Industrial Engineeering Beciyorukler Campus 81620 Duzce-Turkey

Prof. Nazmul Islam

Department of Basic Sciences & Humanities/Chemistry, Techno Global-Balurghat, Mangalpur, Near District Jail P.O: Beltalapark, P.S: Balurghat, Dist.: South Dinajpur, Pin: 733103,India.

Prof. Dr. Ismail Musirin

Centre for Electrical Power Engineering Studies (CEPES), Faculty of Electrical Engineering, Universiti Teknologi Mara, 40450 Shah Alam, Selangor, Malaysia

Prof. Mohamed A. Amr

Nuclear Physic Department, Atomic Energy Authority Cairo 13759, Egypt.

Dr. Armin Shams

Artificial Intelligence Group, Computer Science Department, The University of Manchester.

Editorial Board

Prof. Salah M. El-Sayed

Mathematics. Department of Scientific Computing. Faculty of Computers and Informatics, Benha University. Benha , Egypt.

Dr. Rowdra Ghatak

Associate Professor Electronics and Communication Engineering Dept., National Institute of Technology Durgapur Durgapur West Bengal

Prof. Fong-Gong Wu College of Planning and Design, National Cheng Kung University Taiwan

Dr. Abha Mishra. Senior Research Specialist & Affiliated Faculty. Thailand

Dr. Madad Khan

Head Department of Mathematics COMSATS University of Science and Technology Abbottabad, Pakistan

Prof. Yuan-Shyi Peter Chiu

Department of Industrial Engineering & Management Chaoyang University of Technology Taichung, Taiwan

Dr. M. R. Pahlavani, Head, Department of Nuclear physics, Mazandaran University, Babolsar-Iran

Dr. Subir Das, Department of Applied Mathematics,

Institute of Technology, Banaras Hindu University, Varanasi

Dr. Anna Oleksy

Department of Chemistry University of Gothenburg Gothenburg, Sweden

Prof. Gin-Rong Liu,

Center for Space and Remote Sensing Research National Central University, Chung-Li, Taiwan 32001

Prof. Mohammed H. T. Qari

Department of Structural geology and remote sensing Faculty of Earth Sciences King Abdulaziz UniversityJeddah, Saudi Arabia

Dr. Jyhwen Wang,

Department of Engineering Technology and Industrial Distribution Department of Mechanical Engineering Texas A&M University College Station,

Prof. N. V. Sastry

Department of Chemistry Sardar Patel University Vallabh Vidyanagar Gujarat, India

Dr. Edilson Ferneda

Graduate Program on Knowledge Management and IT, Catholic University of Brasilia, Brazil

Dr. F. H. Chang

Department of Leisure, Recreation and Tourism Management, Tzu Hui Institute of Technology, Pingtung 926, Taiwan (R.O.C.)

Prof. Annapurna P.Patil,

Department of Computer Science and Engineering, M.S. Ramaiah Institute of Technology, Bangalore-54, India.

Dr. Ricardo Martinho

Department of Informatics Engineering, School of Technology and Management, Polytechnic Institute of Leiria, Rua General Norton de Matos, Apartado 4133, 2411-901 Leiria, Portugal.

Dr Driss Miloud

University of mascara / Algeria Laboratory of Sciences and Technology of Water Faculty of Sciences and the Technology Department of Science and Technology Algeria

Prof. Bidyut Saha,

Chemistry Department, Burdwan University, WB, India

International Journal of Physical Sciences

 Table of Contents:
 Volume 11
 Number 20, 30
 October, 2016

ARTICLE

First principles study of the relative stability and electronic properties of the VN/GaN/VN interlayer Miguel Espitia J. R., Jonh Díaz F. and César Ortega L.

262

academicJournals

Vol. 11(20), pp. 262-268, 30 October, 2016 DOI: 10.5897/IJPS2016.4561 Article Number: 5740A3B61495 ISSN 1992 - 1950 Copyright ©2016 Author(s) retain the copyright of this article http://www.academicjournals.org/IJPS

International Journal of Physical Sciences

Full Length Research Paper

First principles study of the relative stability and electronic properties of the VN/GaN/VN interlayer

Miguel Espitia J. R.¹*, Jonh Díaz F.¹ and César Ortega L.²

¹Grupo GEFEM, Universidad Distrital Francisco José de Caldas, Bogotá, Colombia. ²Grupo Avanzado de Materiales y Sistemas Complejos GAMASCO, Departamento de Física, Universidad de Córdoba, Montería Colombia.

Received 8 September, 2016; Accepted 24 October, 2016

Using first principles total-energy calculations within the framework of density functional theory, the relative stability and the structural and electronic properties of a VN/GaN/VN interlayer in sodium chloride (NaCl), cesium chloride (CsCl), nickel arsenide (NiAs), zinc-blende, and wurtzite structures were studied. The calculations were carried out using a method based on pseudopotential, employed exactly as implemented in Quantum-ESPRESSO code. From total energy minimization, it was found out that the global energy minimum of VN/GaN/VN is obtained for the wurtzite structure. Additionally, at high pressure, our calculations show the possibility of phase transition from the wurtzite to the NaCl structure. For the wurtzite phase, the density of states analyses revealed that the interlayer exhibits a half-metallic behavior with a magnetic moment of 2.0 μ_{β} /V-atom. This property essentially comes from the polarization of states V-d and N-p crossing of the Fermi level. Due this property, the interlayer can potentially be used in the field of spintronics as spin injectors.

Key words: Density functional theory (DFT), interlayers, structural and electronic properties.

INTRODUCTION

The group III nitrides such as gallium nitride (GaN) have been the subject of many intense investigations in the last years. This interest is partly motivated by their promising applications for optical devices (Nakamura, 1997) and in high-voltage, high-power, and high-temperature microwave applications (Wagner and Bechstedt, 2002; Nakamura et al., 1994). They are wide-band-gap semiconductors and are used for short-wavelength electroluminescent devices. Additionally, heterostructures based on the surface growth of oxides on GaN are used in various applications, such as high-efficiency blue and green light-emitting devices, ultraviolet light-emitting diodes, metal-oxide-semiconductor field-effect transistors and field-effect transistors (Binari et al., 1997; Ping et al., 1998; Dang et al., 1999). Recently, pioneering theoretical research carried out by Espitia et al. (2015a) has shown that graphene on the GaN(0001) surface is stable and the Dirac cones remain intact.

The ground state of GaN is the wurtzite structure. However, depending on the substrate and the growth

*Corresponding author. E-mail: mespitiar@udistrital.edu.co.

Author(s) agree that this article remain permanently open access under the terms of the <u>Creative Commons Attribution</u> <u>License 4.0 International License</u> conditions, a zinc-blende phase can be formed (Yeh et al., 1992). The difference in total energy between the zinc-blende and wurtzite structures is very small (~10 meV/atom), which explains why the two structures can be obtained experimentally. Additionally, this last year, there has been a great deal of interest in developing new and improved ultrahard materials using the group III nitrides and a metal transition ion. Experimental studies show that these superhard compounds can be grown; for example, a TiN/GaN multilayer was grown via reactive pulsed laser deposition (Rawat and Sands, 2006), while the AIN/VN superlattice was grown via reactive sputtering (Kim et al., 2001). On the other hand, the combination of the semiconductor properties of GaN and the magnetic properties of the transition metal ions is of great current interest, because of their potential applications in diluted magnetic semiconductors (DMS) (Yao et al., 2012; Sato et al., 2013; Cudris et al., 2016). The study of such ultrahard structures or those with magnetic properties is clearly of high technological relevance, but it is also fundamentally interesting with regard to understanding the mechanisms responsible for the extreme hardness and the magnetic properties. Achieving super hardness or specific magnetic properties involves the ability to deliberately create stable and metastable structures according to one's desire on an atomic level. The realization of this objective is becoming increasingly achievable due to advancement in experimental techniques as well as to theoretical methods that are able to reliably predict the relative stability and the atomic and electronic properties of increasingly complex structures.

For this reason, in this paper, a detailed theoretical calculation of the structural and electronic properties of VN/GaN/VN was reported in order to determine the energetically most favorable structure. To his end, the results of the calculated structural and magnetic properties in the wurtzite, zinc-blende, rock-salt, cesium chloride (CsCI), and nickel arsenide (NiAs) structures at zero temperature were presented.

The organization of the paper is as follows: section 2 presents a description of the pseudopotential method used, with some computational details. Section 3 presents the calculated results, with discussion of the structural properties. The transition phase was studied and the energy formation was calculated, and for the most stable structure, the energetically most favorable magnetic phase was calculated. Finally, the electronic properties of the wurtzite structure were studied. In section 4, some conclusions were drawn.

COMPUTATIONAL METHODS

The calculations were performed within the density functional theory (DFT) framework using the Quantum-ESPRESSO package (Giannozzi et al., 2009). The correlation and exchange effects of the electrons were treated using the generalized gradient approximation (GGA) of Perdew, Burke, and Ernzerhof (PBE) (Perdew et al., 1996). Electron-ion interactions were treated with

the pseudopotential method (Vanderbilt, 1990; Laasonen et al., 1993). The electron wave functions were expanded into plane waves with a kinetic-energy cutoff of 40 Ry. For the charge density, a kinetic energy cutoff of 400 Ry was used. A $6 \times 6 \times 4$ Monkhorst-Pack mesh (Monkhorst and Pack, 1976) was used to generate the k-points in the unit cell. The calculations were performed taking into account the spin polarization.

To calculate the lattice constant, the minimum volume, the bulk modulus, and the cohesive energy of the two structures studied, calculations were adjusted to the Murnaghan equation of state (Murnaghan, 1944):

$$E(V) = E_0 + \frac{B_0 V}{B'_0} \left[\frac{\binom{V_0}{V}^{B'}}{B'_0 - 1} + 1 \right] - \frac{B_0 V_0}{B'_0 - 1}$$
(1)

where B_0 is the bulk modulus, its first derivative is B'_0 , V_0 is the equilibrium volume of the cell, and E_0 represents the cohesive energy.

The electronic configurations used in the present calculations are [Ar] $3d^{10}4s^24p^1$ for the Ga, [Ar] $3d^34s^2$ for the V, and [He] $2s^22p^3$ for the N. In each of the five phases, the interlayer is modeled as sandwiching two layers of GaN between two layers VN.

RESULTS AND DISCUSSION

Structural properties

To determine the structural properties in the ground state, such as the lattice constant (a), the bulk modulus (B_0), the c/a ratio, and the total energy (E_0) of the VN/GaN/VN interlayer in the CsCl, NaCl, NiAs, zinc-blende, and wurtzite structures, the total energy was calculated as a function of the volume, and the results were fit to the Murnaghan equation of state. Figure 1 shows the energy-volume curves.

In Figure 1, it can be noted that the five structures considered in this study are metastable, because in each of the energy-volume curves there is a minimum energy value. In Figure 1, one can clearly see that the wurtzite phase is the energetically most favorable structure, because it has the lowest minimum energy value. Additionally, we can see that the curve corresponding to the wurtzite phase crosses the curve of the NaCI structure of smallest equilibrium volume. This indicates a high-pressure phase transition from the wurtzite to the NaCl structure. To analyze the phase transformation under high pressure, we used the Gibbs free energy, given by Mujica et al. (2003): G = U + PV - TS; where U is the internal energy, P is the pressure, V is the volume, T is the temperature, and S is the entropy. Since the calculations are in ground state (T = 0 degree K), the last term of the Gibbs energy can be neglected, and by working with the enthalpy, the Gibbs free energy reduces to H = U + PV (Mancera et al., 2004; Espitia et al., 2015b). This equation was used for the five crystal structures considered in this paper; however, the possible existence of other stable or metastable structures for a VN/GaN/VN interlayer cannot be excluded.



Figure 1. Total energy as a function of unit-cell volume for the CsCl, NaCl, wurtzite, NiAs, and zinc-blende structures of VN/GaN/V interlayer.



Figure 2. Enthalpy as a function of pressure in the NaCl and wurtzite phases of VN/GaN/VN interlayer.

Figure 2 shows enthalpy as a function of pressure for the NaCl and wurtzite phases. It was noted that the compound will transform from a wurtzite structure to NaCl at a transition pressure $P_T = 21.7$ GPa. Note that before the transition ($P < P_T$), the lowest values of enthalpy correspond to the wurtzite phase. This shows the relative stability of the wurtzite phase within this pressure range, while for higher values of the transition pressure ($P > P_T$), the NaCl phase has the lowest values of enthalpy. Therefore, within this pressure range, NaCl is the most stable phase. At the transition pressure, the enthalpies of both structures are equal. Across the transition, there is a volume reduction of about 14.3%, from 92.122 Å³ to 78.07 Å³. Similar transitions, from the wurtzite to the NaCl structures, have been observed experimentally and predicted theoretically for GaN (Arbouche et al., 2009; Perlin et al., 1992), and they have been theoretically predicted in other compounds (López et al., 2008; Báez et al., 2013).

The lattice constant a_0 , the c/a value, the bulk modulus (B₀), and the total energy (E₀) of the binary compounds GaN and VN calculated in the ground state, wurtzite for GaN and NaCl for VN, are shown in Table 1, while Table 2 shows the structural parameters of VN/GaN/VN calculated in the phases CsCl, NaCl, NiAs, wurtzite, and zinc-blende.

Table 1 compares the experimental and theoretical data available from the literature. We can see that the

Compound	Phase	<i>a</i> ₀ (Å)	c/a	B₀ (GPa)	E₀ (eV)
		3.205	1.64	175.33	
GaN	Wurtzite	3.22 ^a	1.634 ^a	176.54 ^a	-2513.12
		3.19 ^b	-	188 ^b	
		4.10	-	228	
VN	NaCl	4.06 ^c	-	233 ^d	-2409.48
		4.14 ^e	-	-	

Table 1. Lattice constant, c/a ratio, bulk modulus, and total energy of the binary compounds GaN and VN.

^aTheoretical Reference (Arbouche et al., 2009), ^bExperimental Reference (Shultz and Thiemann, 1977), ^cTheoretical Reference (Stampfl and Freeman, 2012), ^dExperimental Reference (Stampfl et al., 2001). ^cTheoretical Reference (Carlson et al., 1986).

Table 2. Lattice constant, c/a ratio, equilibrium volume, bulk modulus, total energy, and magnetic moment per cell of the VN/GaN/VN interlayer.

Phase	<i>a</i> ₀ (Å)	c/a	V ₀ (Å ³)	B₀ (GPa)	E ₀ (eV)
CsCl	2.648	-	72.120	228.060	-2458.323
NaCl	2.923	1.414	78.070	211.794	-2459.697
NiAs	2.864	1.81	74.157	247.487	-2459.622
Wurtzite	3.193	1.64	92.122	161.746	-2460.201
Zinc-blende	4.505	-	96.966	177.940	-2460.035

equilibrium lattice constant a_0 of the binary compounds GaN and VN calculated in this paper is in very good agreement with theoretical references (Arbouche et al., 2009; Stampfl and Freeman, 2012; Carlson et al., 1986) and experimental references (Shultz and Thiemann, 1977; Stampfl et al., 2001), since it differs by less than one percent. On other the hand, despite the difference between the crystal structures wurtzite, GaN, and NaCl VN, experimental and theoretical studies show that the GaN:V system can be grown. Souissi et al. (2005) report on the vanadium doping of a GaN layer grown using the metal-organic vapor phase epitaxy technique (MOVPE), and González et al. (2007) studied the structural properties of Ga_{1-x}V_xN using DFT.

In Table 2, it can see be seen that the phase with the lowest value of energy corresponds to the wurtzite structure. Additionally, it can see also been seen that the values of the bulk modules of the VN/GaN/VN interlayer are higher, which confirms that they are quite rigid, making them good candidates for possible applications in devices operated at high temperature and high power, as well as in hard coatings.

In order to verify the relative stability of the VN/GaN/VN interlayer with reference to the terminal phases, the corresponding formation energy was calculated, which is expressed as the difference between the total energy of the ternary VN/GaN/VN phases, $E_{VN/GaN/VN}^{phase}$ and the reference states of wurtzite GaN, $E_{GaN}^{wurtzite}$ and NaCl of

VN, E_{VN}^{NaCl} (Zhang and Veprek; Vargas-Hernández et al., 2015; Ortega et al., 2016):

$$E_{f} = E_{VN/GaN/VN}^{phase} - x E_{GaN}^{wurtzita} - (1 - x) E_{VN}^{NaCl}$$

For all interlayers x = 0.5, because there are the same number of molecules of GaN and VN. Table 3 shows the values of formation energy E_f calculated in each of the five phases considered in this study.

According to the results shown in Table 3, the smallest value of energy of formation that corresponds to the VN/GaN/VN interlayer is the wurtzite phase; therefore, it is the most energetically stable structure. This confirms the result obtained earlier. Additionally, the moderate formation energy values indicate that the compounds can easily be grown experimentally.

These results for the energy of formation are important, because knowing these values, growing conditions can be improved and therefore a VN/GaN/VN interlayer of excellent quality can be grown.

For the most stable phase, the wurtzite structure, the total energy variation as a function of the volume of the ferromagnetic (FM) and antiferromagnetic (AFM) phases were calculated in order to determine the most favorable magnetic phase of the interlayer. Figure 3 shows the energy-volume curve for the VN/GaN/VN interlayer in the wurtzite structure in the FM and AFM states. In the ground state, the total energy difference between the FM

Table 3. Formation energy.

Compound	Phase	E _f (eV)
	CsCl	2.977
	NaCl	1.603
VN/GaN/VN	NiAS	1.678
	Wurtzite	1.099
	Zinc-blende	1.265



Figure 3. Total energy as a function of unit-cell volume for the wurtzite structures of VN/GaN/VN interlayer in the FM and AFM phases.

and AFM states ($\Delta E = E_{FM} - E_{AFM}$) was - 0.03 eV. The FM state was more energetically favorable than the AFM state.

Electronic properties

The theoretical lattice constants a_0 and the c/a ratio of the interlayer in the most energetically favorable phase, the wurtzite structure, shown in Table 1, were used to calculate the spin-polarized density of states (DOS) along the high-symmetry direction in the first Brillouin zone.

Figure 4 shows the total and partial spin-polarized density of states (TDOS and PDOS, respectively) of the VN/GaN/VN interlayer in the wurtzite structure. The TDOS results show that the interlayer is half-metallic and ferromagnetic. This result occurs because in the valence band near the Fermi level, the majority spins (spin-up) are metallic, and the minority spins (spin-down) are semiconductors. This interlayer has a spin polarization of 100% of the conduction carriers in the ground state, which is a requirement for spin injectors (Vargas-Hernández et al., 2015; Espitia et al., 2016). This finding suggests that the interlayer can be efficiently used as a spin injector. Figure 4 shows that in the valence band near the Fermi level, the spin-up density (the majority

spins) is mainly dominated by the V-3d states and to a lesser extent by the N-2p states, which cross the Fermi level. The interlayer has a magnetic moment of 4 μ_{β} per supercell. Since there are two vanadium atoms in the supercell, this corresponds to 2 μ_{β} /V-atom. The magnetic moment is due to the V³⁺ configuration (with electronic configuration $[Ar]3d^2$). This implies that when the V atom is in the interlayer, it gives up three electrons, and two valence electrons remain (d² configuration). These valence electrons couple ferromagnetically, and as a result the two electrons produce a total magnetic moment of 2 μ_{β} /atom-V. Touati et al. (2008) reported a valence of V³⁺ for infrared luminescence in V-doped GaN samples grown with MOVPE on a sapphire substrate. The calculated magnetic moment for VN/GaN/VN in the wurtzite structure is an integer; therefore, this confirms that the interlayer is ferromagnetic and half-metallic.

Conclusions

The first principles calculations to determine the structural, electronic, and magnetic properties of a VN/GaN/VN interlayer for CsCl, NaCl, NiAs, zinc-blende, and wurtzite structures, within the framework of density functional theory with the approach of the generalized



Figure 4. Partial and total density of states for VN/GaN/VN wurtzite structure.

gradient approximation (GGA) were reported. From the computation of the total energy versus unit-cell volume, the wurtzite structure was found to be preferred at ambient pressure. It was found out that the transition pressure for the wurtzite to the NaCl phase was about 21.7 GPa. The interlayer has a half-metallic behavior with a magnetic moment of 2 μ_{β} /V-atom. The ground-state ferromagnetic behavior comes from the polarization of the V-3d and N-2p orbitals, which cross the Femi level. These compounds are good candidates for potential applications in spintronics and can be used as spin injectors.

Conflict of Interests

The authors have not declared any conflict of interests.

ACKNOWLEDGEMENTS

The authors thank the Research Center of the Distrital University CUID for its financial support.

REFERENCES

- Arbouche O, Belgoumène B, Soudini B, Driz M. (2009). First principles study of the relative stability and the electronic properties of GaN, Computational Materials. Science. 47:432-438.
- Báez R, Ortega C, Espitia M. (2013). First-Principles Study of Structural and Electronic Properties of Chromium Nitride/Gallium Nitride (CrN/GaN) Multilayer. Rev. Ingeniería Cienc. 9:163-174.
- Binari SC, Kruppa W, Dietrich HB, Kelner G, Wickenden AE, Freitas JA (1997). Fabrication and characterization of GaN FETs. Solid- State Electron. 41:1549-1554.
- Carlson ON, Smith JF, Nafziger RH (1986). The vanadium-nitrogen system: a review. Metallurgical Mater. Trans. A. 17:1647-1656.

- Cudris E, Díaz J, Espitia M (2016). Relative phase and physical properties of CrN/AIN multilayer: A DFT study. J. Phys. Conf. Series 743(2016):012003.
- Dang XZ, Asbeck PM, Yu ET, Sullivan GJ, Chen MY, McDermot BT, Boutros KS, Redwing JM (1999). Measurement of drift mobility in AlGaN/GaN heterostructure field-effect transistor. Appl. Phys. Lett. 74:3890.
- Espitia M, Díaz J, Ortega C (2016). Ab initio calculations of half-metallic ferromagnetims behavior of (VN)₁/(AlN)₁, (VN)₁/(GaN)₁, and (MnN)₁/(InN)₁. J. Phys. Conf. Series. 687:012069.
- Espitia M, Díaz -Forero J, Castillo-Méndez L (2015a). First-principles calculations of the pressure dependence on the structural and electronic properties GaN/CrN superlattice. Rev. Fac. de Ing. Uni. Antioquia. 76:143-147.
- Espitia-Rico M, Rodríguez-Martínez J, Moreno-Armenta G, Takeuchi N (2015b). Graphene monolayers on GaN(0001). Appl. Surf. Sci. 326:7-11.
- Giannozzi P, Baroni S, Bonin N (2009). QUANTUM ESPRESSO: a modular and open-source software project for quantum simulations of materials. J. Phys. Condens. Matter 21:395502.
- González R, López W, Rodríguez JA (2007). First-principles calculations of structural properties of GaN:V. Solid State Commun. 144:109-113.
- Kim IW, Li Q, Marks LD, Barnett SA (2001). Critical thickness for transformation of epitaxially stabilized cubic AIN in superlattices. Appl. Phys. Lett. 78:892.
- Laasonen K, Pasquarello A, Car R, Lee C, Vanderbilt D (1993). Car-Parrinello molecular dynamics with Vanderbilt ultrasoft pseudopotentials. Phys. Rev. B. 47:10142.
- López W, Rodríguez JA, González R, Fajardo F, Mancera L (2008). Pressure effect on electronic properties of Sc0.5In0.5N compound. Phys. stat. sol. B. 246:1-5.
- Mancera L, Rodríguez JA, Takeuchi N (2004). Theoretical study of the stability of wurtzite, zinc-blende, NaCl and CsCl phases in group IIIB and IIIA nitrides. Phys. Status Solid B. 241.
- Monkhorst H, Pack J (1976). Special points for Brillouin-zone integrations. Phys. Rev. B. 13:5188.
- Mujica A, Rubio A, Muñoz A, Needs RJ (2003). High-pressure phases of group-IV, III–V, and II–VI compounds. Rev. Modern Phys. 75:868.
- Murnaghan FD (1944). The Compressibility of media under pressure. Proc. Nat. Acad. Sci. U.S.A. 30:244-247.
- Nakamura S (1997). III-IV Nitride based light-emitting devices. Solid State Commun. 102(2-3):237-243.
- Nakamura S, Mukai T, Senoh M (1994). Candela-class high-brightness

InGaN/AlGaN double-heterostructure blue-light-emitting diodes. Appl. Phys. Lett. 64(13):1687-1692.

- Ortega C, Casiano J, Espitia M (2016). Electronic and magnetic properties GaN/MnN/GaN and MnN/GaN/MnN interlayers. J. Phys. Conf. Series 687(2016):012052.
- Perdew J, Burke K, Ernzerhof M (1996). Generalized Gradient Approximation Made Simple. Phys. Rev. Lett. 77:3865.
- Perlin P, Jauberthie-Carillon C, Lti JP, Miguel AS, Grzegory I, Polian A (1992). Raman scattering and x-ray-absorption spectroscopy in gallium nitride under high pressure. Phys. Rev. B 45:83.
- Ping AT, Chen Q, Yang JW, Asif Khan M, Adesida I (1998). DC and Microwave Performance of High-Current AlGaN/GaN Heterostructure Field Effect Transistors Grown on p-Type SiC Substrates. IEEE Electron. Dev. Lett. 19:54-56.
- Rawat V, Sands T (2006). Growth of TiN/GaN metal/semiconductor multilayers by reactive pulsed laser deposition. J. Appl. Phys. 100:064901.
- Sato K, Dederics PH, Yoshida K (2013). curie temperatures of III-V diluted magnetic semiconductors calculated from first principles. Europhys. Lett. 61(403):00191.
- Shultz H, Thiemann KH (1977). Crystal structure refinement of AIN and GaN. Solid State Commun. 23:851-819.
- Souissi M, Bchetnia A, El Jani B (2005). Growth of vanadium-doped GaN by MOVPE. J. Cryst. Growth. 277:57-63.
- Stampfl C, Freeman AJ, Mannstadt W, Asahi R (2001). Electronic structure and physical properties of early transition metal mononitrides: Density-functional theory LDA, GGA, and screenedexchange LDA FLAPW calculations. Phys. Rev. B. 63:155106.
- Stampfl C, Freeman AJ (2012). Structure and stability of transition metal nitride interfaces from first-principles: AIN/VN, AIN/TiN, and VN/TiN. Appl. Surf. Sci. 258:5638-5645.
- Touati H, Souissi M, Chine Z, El Jani B (2008). Near-infrared photoluminescence of V-doped GaN. Microelectron. J. 39:1457-1460.
- Vanderbilt D (1990). Soft self-consistent pseudopotentials in a generalized eigenvalue formalism. Phys. Rev. B. 41:7892.

- Vargas-Hernández C, Espitia M, Báez R (2015). Half-metallic ferromagnetism of Zn_xMn_{1-x}O compounds: A first-principles study. Comput. Condens. Matter. 4:1-5.
- Wagner JM, Bechstedt F (2002). Properties of strained wurtzite GaN and AIN: Ab initio studies. Phys. Rev. B. 66:115202.
- Yao G, Fan G, Zheng S, Ma J, Chen J, Zhou D, Li S, Zhang Y, Su S (2012). First-principles analysis on V-doped GaN. Opt. Mater. 34:1593-1597.
- Yeh CY, Lu ZW, Froyen S, Zunger A (1992). Zinc-blende–wurtzite polytypism in semiconductors. Phys. Rev. B. 46:10086.
- Zhang RF, Veprek S (2007). Phase stabilities and spinodal decomposition in the Cr1–xAlxN system studied by ab initio LDA and thermodynamic modeling: Comparison with the Ti1–xAlxN and TiN/Si3N4 systems. Acta Mater. 55:4615-4624.

International Journal of Physical Sciences

Related Journals Published by Academic Journals

African Journal of Pure and Applied Chemistry
 Journal of Internet and Information Systems
 Journal of Geology and Mining Research
 Journal of Oceanography and Marine Science
 Journal of Environmental Chemistry and Ecotoxicology
 Journal of Petroleum Technology and Alternative Fuels

academiclournals